

Conversion factors for soil test results from laboratories using the Bray P1, Ammonium Acetate (AmmAc), and 0.1 N HCl extracts.

| Nutrient | Reported Units | Conversion Equation |
|----------|----------------|--|
| P | ppm P | $UD\text{-FIV} = (1.23 \times \text{Bray P1-P}) + 6.07$ |
| | lb P/ac | $UD\text{-FIV} = (0.61 \times \text{Bray P1-P}) + 6.07)$ |
| K | ppm K | $UD\text{-FIV} = (0.55 \times \text{AmmAc-K}) - 0.31$ |
| | lb K/ac | $UD\text{-FIV} = (0.27 \times \text{AmmAc-K}) - 0.31$ |
| Ca | ppm Ca | $UD\text{-FIV} = (0.12 \times \text{AmmAc-Ca}) - 5.60$ |
| | lb Ca/ac | $UD\text{-FIV} = (0.06 \times \text{AmmAc-Ca}) - 5.60$ |
| Mg | ppm Mg | $UD\text{-FIV} = (0.84 \times \text{AmmAc-Mg}) - 0.13$ |
| | lb Mg/ac | $UD\text{-FIV} = (0.42 \times \text{AmmAc-Mg}) - 0.13$ |
| Mn | ppm Mn | Mehlich 3 Mn (lb/ac) = $(1.45 \times \text{HCl-Mn}) + 14.61$ |
| | lb Mn/ac | Mehlich 3 Mn (lb/ac) = $(0.73 \times \text{HCl-Mn}) + 14.61$ |
| Zn | ppm Zn | Mehlich 3 Zn (lb/ac) = $(1.91 \times \text{HCl-Zn}) + 1.22$ |
| | lb Zn/ac | Mehlich 3 Zn (lb/ac) = $(0.96 \times \text{HCl-Zn}) + 1.22$ |

Example: Converting Bray P1 (P) soil test results to UD-FIV

Soil Test Results: Bray P1-P = 75 lb ppm P

Use the equation for results reported in ppm: $UD\text{-FIV} = (1.23 \times \text{Bray P1-P}) + 6.07$

$$UD\text{-FIV} = (1.23 \times 75 \text{ ppm P}) + 6.07 = 98 \text{ FIV}$$