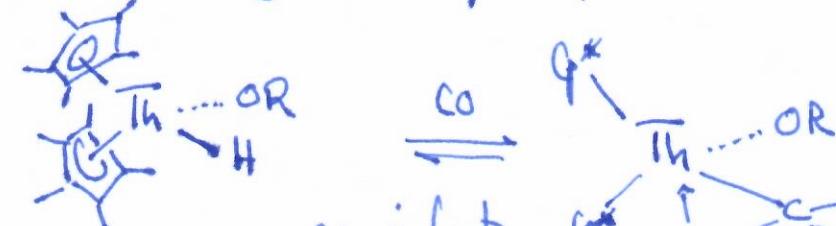


Insertion/Elimination cont'd

Lecture 15

T. Marks JACS 1981, 103, 6959



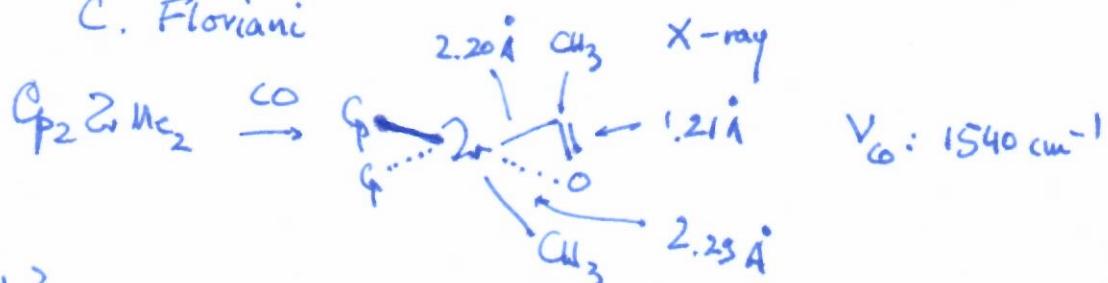
$R = C^+ Bu_2H$
eq. in fast
on 'H-NMR time
Scale'

'oxycarbene' resonance form
 $\nu_{CO} : 1477 \text{ cm}^{-1}$
 $\nu_{CH} : 372 \text{ ppm}$ $J_{CH} : 114 \text{ Hz}$
 $\delta_{CH} : 15.2 \text{ ppm}$

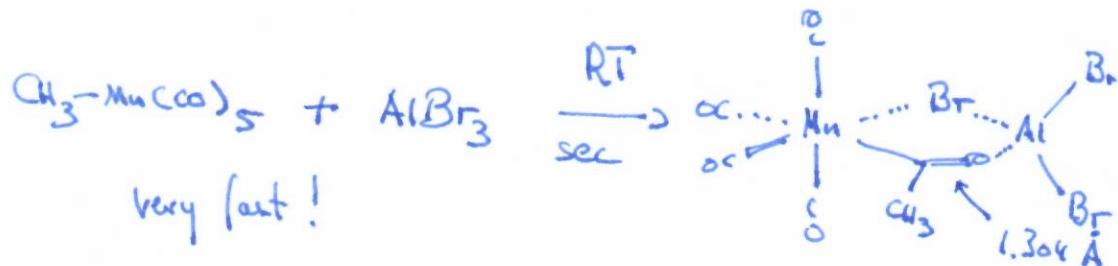
eq. thermodynamics:

$\Delta H = -4.5 \text{ kcal/mol}$ due to added oxophilic interaction
of actinide
 $\Delta S = -11.7 \text{ e.u.}$

other 'n²-acngls' C. Floriani

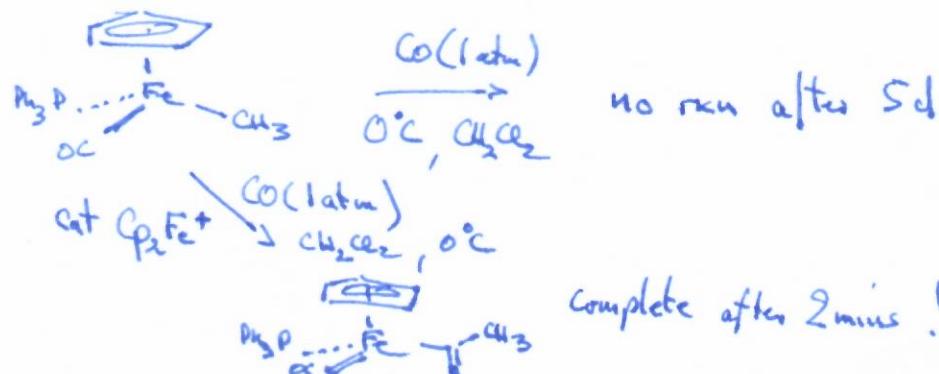


external Lewis Acid?



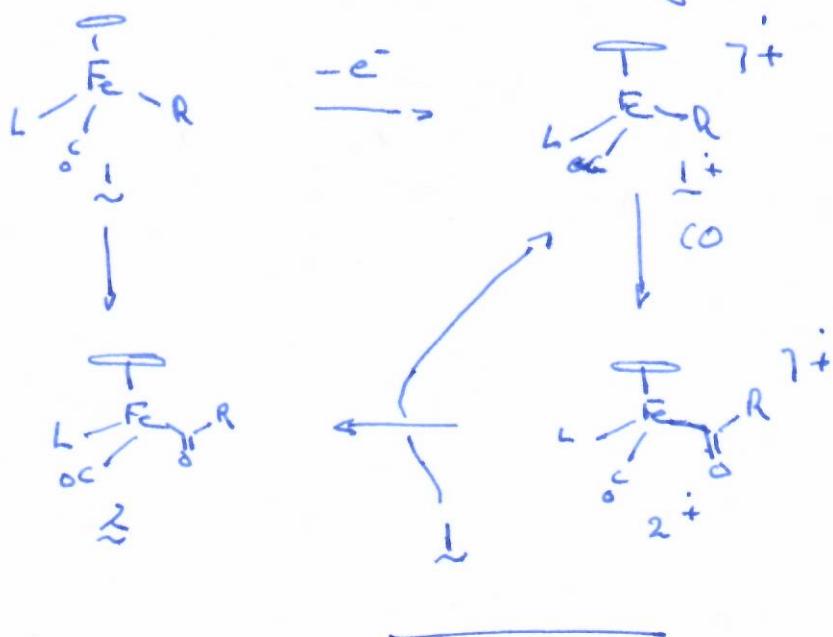
'Oxidatively induced CO-insertion'

R. Magnusson OM 1983, 2, 460



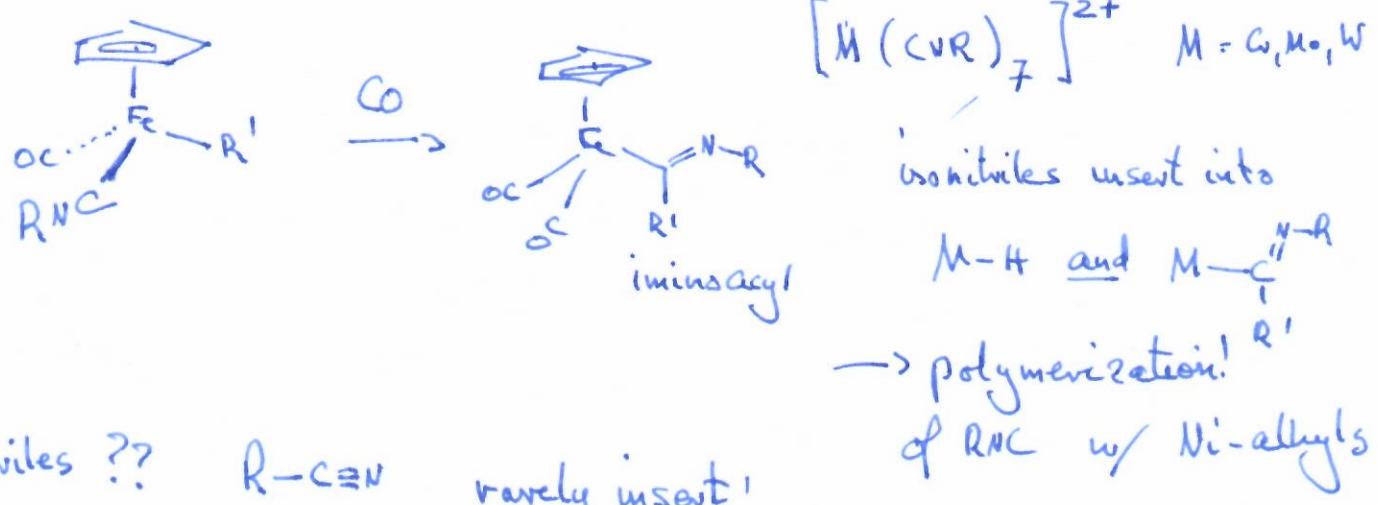
Mechanism? ; electron transfer catalysis (ETC)

2

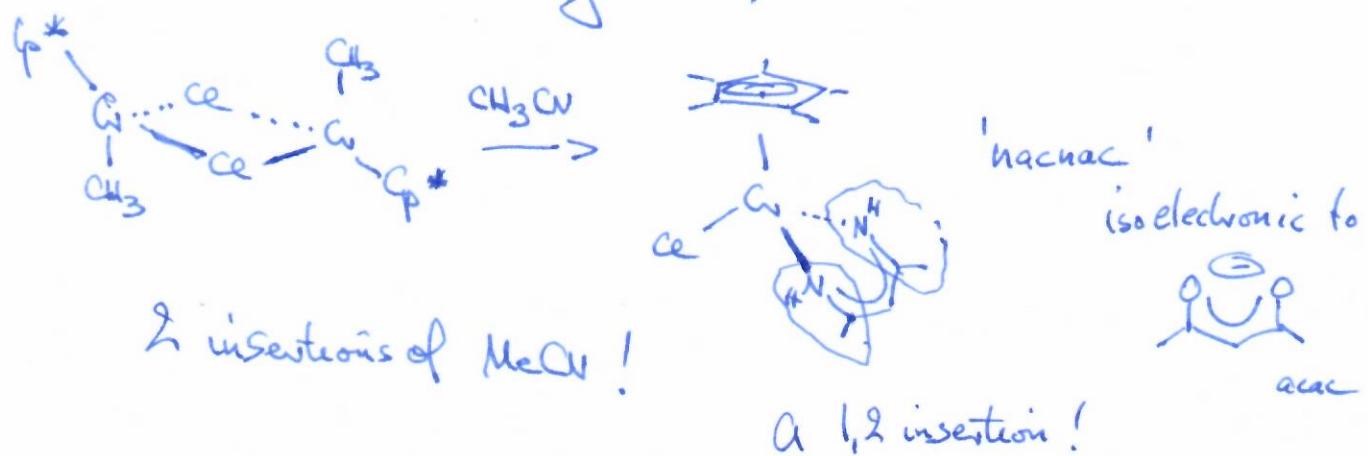


other 1,1 insertions

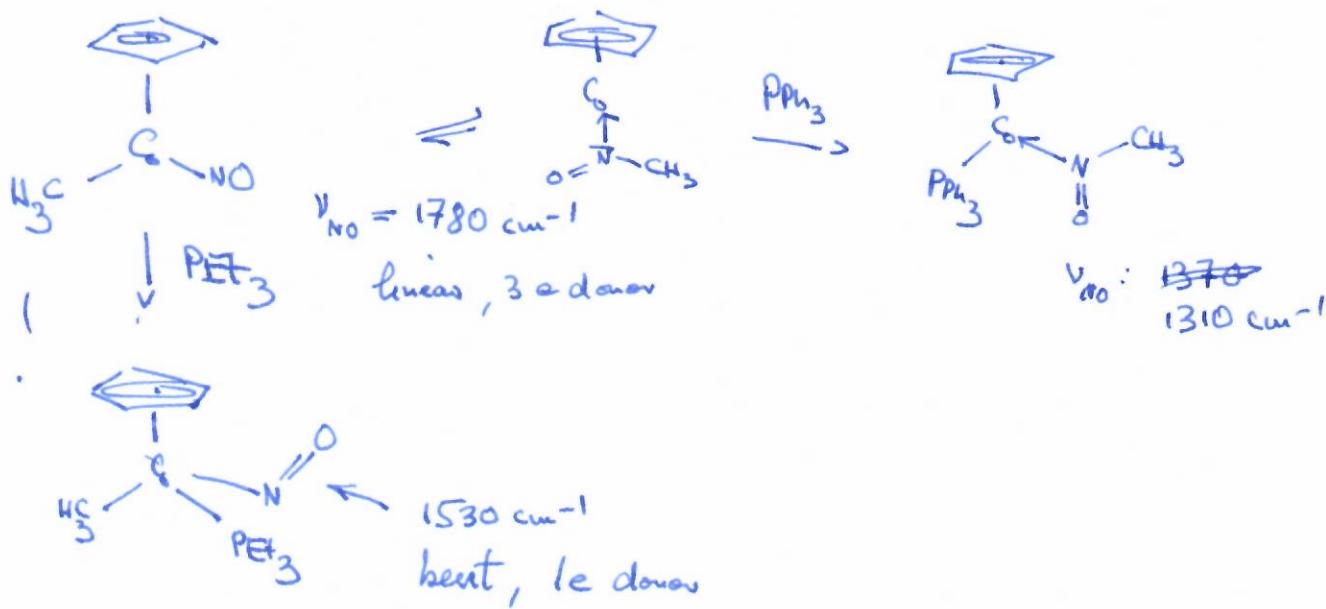
isonitriles : $\text{R}-\text{N}\equiv\text{C}$ ($\text{O}\equiv\text{C}$) better σ -donors } than CO
(isocyanides) weaker π -acceptors }



? nitriles ?? $\text{R}-\text{C}\equiv\text{N}$ rarely insert!



nitrosyl: NO

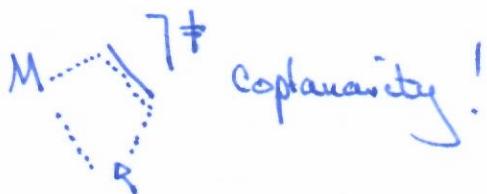


1,2 insertions : alkene insertion



$\text{R} = \text{H}$ microscopic reverse
of β -hydrogen elimination
very fast, reversible

$\text{R} = \text{alkyl, aryl} \rightarrow$ olefin polymerization
microscopic reverse:
 β -alkyl elimination, rare
but exists!



Selective trimerization of //

