Graduate Student Paper Competition

Investigation of the Residence Time Effect on Phosphate (P) Adsorption/Desorption

Mechanisms at the Ferrihydrite/Water Interface using P XANES and ATR-FTIR
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The fate and transport of P in acid soils is highly dependent on adsorption/desorption reactions at the amorphous mineral/water interface. In this study, we investigated residence time (<24h and >24h) effects on P adsorption/desorption at the ferrihydrite/water interface at pH=4, I=0.1M NaCl. and [P] = 1mM. Adsorption kinetics were initially fast, resulting in 93% of the total adsorption within 6h, followed by a slow P uptake. Desorption were carried for 24h under same reaction conditions using a stirred-flow apparatus. Only ≥5% of total adsorbed P is recovered after being incubated for 2d and the total P recovery decreases to <1% with increasing aging time from 2d to10months. Temperature studies show that the overall adsorption is an endothermic reaction, suggesting a predominantly chemically-controlled reaction. In-situ FTIR studies suggested the formation of inner-sphere (monodentate mononuclear or bidentate binuclear) complexes and whereas P-XANES indicated the formation of an amorphous iron phosphate-like surface precipitates.

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