

Monday, 6 October 2008

Individual/Group Quiz on Photosynthesis Case Study Problem, Are you what you eat?

(To do well on this quiz you should be familiar with photosynthesis, Calvin Cycle, radioisotopic tracers, phase plane plots, light and dark reactions of photosynthesis, and the Bassham experiment.)

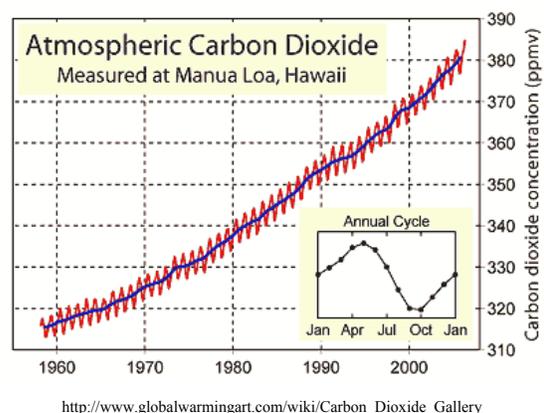
Select the *best* answer.

- \_\_\_\_ 1. The Calvin Cycle is composed of photosynthetic
- “Dark” reactions that normally operate only at night.
  - “Dark” reactions that normally operate only in day light.
  - “Light” reactions that normally operate only at night.
  - “Light” reactions that normally operate only in day light.

- \_\_\_\_ 2. The ash that remains after wood is burned.
- Is composed primarily of carbon
  - Represents salts and minerals extracted from the soil
  - Originally came from the air through the leaves of a tree.
  - Comes mostly from decomposed chlorophyll

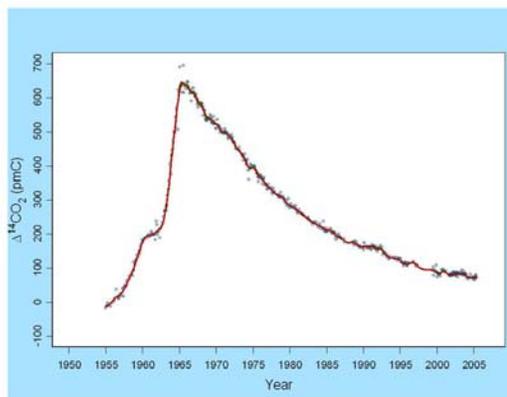
- \_\_\_\_ 3. Plants
- Produce both CO<sub>2</sub> and O<sub>2</sub> in the day, but not at night.
  - Consume both CO<sub>2</sub> and O<sub>2</sub> at night, but not in the day.
  - Produce O<sub>2</sub> during the day and consume CO<sub>2</sub> at night.
  - Consume CO<sub>2</sub> in the day and consume O<sub>2</sub> at night.

- \_\_\_\_ 4. The “Keeling Curve” (right) tracks atmospheric CO<sub>2</sub> concentrations over time. From this curve one can deduce:
- Manua Loa is probably in the Southern Hemisphere.
  - The efficiency of photosynthesis has likely improved ~15% since 1960.
  - Global warming has caused an increase in CO<sub>2</sub> concentrations.
  - Globally, biological respiration is out pacing photosynthesis.



- \_\_\_\_ 5. The carbon atoms in nylon
- Would have a significantly different  $\delta^{13}\text{C}$  than gasoline.
  - Are derived directly by chemical synthesis from atmospheric CO<sub>2</sub>.
  - Are derived mostly from corn starch and cellulose.
  - Include essentially no <sup>14</sup>C.

6. The curve at the right displays the amount of  $^{14}\text{C}$  remaining in the atmosphere as the result of atmospheric atomic bomb testing in the 1950s and 60s. If a 100 year old tree were cut down in 2005 and the  $^{14}\text{C}$  content of each annual growth ring was measured and plotted, the resulting graph of  $^{14}\text{C}$  vs time would:



- A. Look much like the graph at the right.
- B. Look like a saturation curve with highest values in 2005.
- C. Be flat corresponding to the highest point in 1968.
- D. Be flat corresponding to 2005 levels.

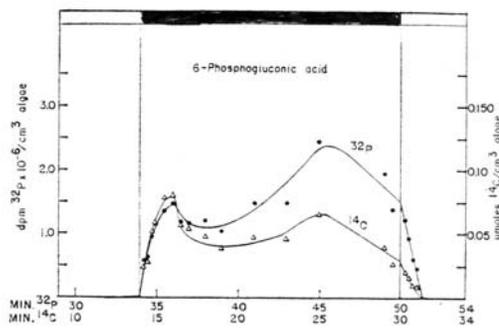
7. The *primary* purpose of the Bassham experiment was: (Pick the *best* answer.)

- A. To study the regulation of carbon flux in the Calvin Cycle.
- B. To study the incorporation of radioisotopes into Calvin Cycle intermediates.
- C. To study the effect of light on 3PGA concentrations.
- D. To study the dynamics of  $^{32}\text{P}$  and  $^{14}\text{C}$  in the dark reactions of photosynthesis.

8. The belemnite used as a  $\delta^{13}\text{C}$  reference standard and also the state fossil of Delaware

- A. Has a lower  $^{13}\text{C}/^{12}\text{C}$  ratio than the carbon in either  $\text{C}_3$  or  $\text{C}_4$  plants.
- B. Is composed of hydroxyapatite,  $\text{Ca}_5(\text{PO}_4)_3\text{OH}$ .
- C. Is an extinct type of mollusk related to squid and cuttlefish.
- D. Has a  $^{14}\text{C}/^{12}\text{C}$  ratio similar to that in atmospheric  $\text{CO}_2$ .

9. The curve at the right displays the amount of  $^{14}\text{C}$  and  $^{32}\text{P}$  in (PGNA) 6-phosphogluconate in Bassham's experiment. It shows that:



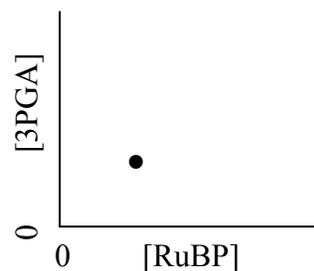
- A. After 15 min of darkness  $\geq 60\%$  of the carbon in PGNA comes from starch.
- B. The oxidative part of the pentose phosphate pathway operates only in the dark.
- C. PGNA is not part of the Calvin Cycle.
- D. All of the above.

10. Ignoring photorespiration, the ratio of ATP:NADPH consumed per each  $\text{CO}_2$  fixed by the Calvin Cycle is \_\_\_ in  $\text{C}_3$  plants:?

- A. 2:1
- B. 3:2
- C. 1:1
- D. 1:2

11. Consider two identical photosynthesizing leaves placed in separate vessels containing atmospheric carbon dioxide. Vessel 1 is sealed so new air cannot come in, while the Vessel 2 remains open to the air. Photosynthesis is allowed to continue until the  $\text{CO}_2$  in Vessel 1 is used up. What can you predict about the  $\delta^{13}\text{C}$  values for newly fixed carbon?
- A.  $\delta^{13}\text{C}$  atmospheric  $\text{CO}_2 = \delta^{13}\text{C}$  Vessel 1 <  $\delta^{13}\text{C}$  Vessel 2
  - B.  $\delta^{13}\text{C}$  atmospheric  $\text{CO}_2 < \delta^{13}\text{C}$  Vessel 1 <  $\delta^{13}\text{C}$  Vessel 2
  - C.  $\delta^{13}\text{C}$  atmospheric  $\text{CO}_2 = \delta^{13}\text{C}$  Vessel 1 >  $\delta^{13}\text{C}$  Vessel 2
  - D.  $\delta^{13}\text{C}$  atmospheric  $\text{CO}_2 > \delta^{13}\text{C}$  Vessel 1 >  $\delta^{13}\text{C}$  Vessel 2 .

12. The point on the graph to the right represents the concentrations of 3PGA and RuBP at the moment the lights were turned off in Bassham's experiment. Plot the point corresponding to their concentrations about 30 sec later when [3PGA] peaks. What would be the expected slope ( $m = \Delta[3\text{PGA}]/\Delta[\text{RuBP}]$ ) of the line connecting ● and the point you plotted?



- A.  $m \leq -1$
- B.  $-1 < m < 0$
- C.  $0 < m < 1$
- D.  $m \geq 1$