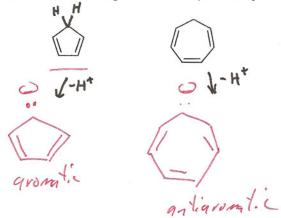
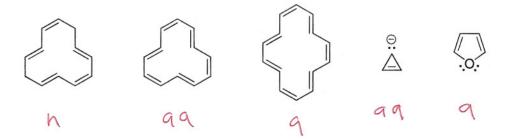
1. Which of the molecules below is the stronger acid? Explain why.

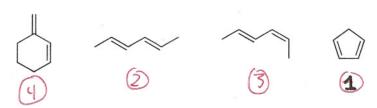


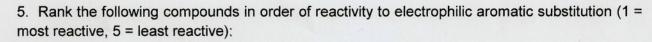
2. Label each of the following molecules as being aromatic, antiaromatic, or neither:

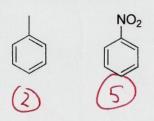


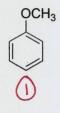
3. Multiple choice: which molecule below is capable of being chiral? Circle the one correct answer.

4. Number the following dienes in order of Diels-Alder reactivity (1 = most reactive diene, 4 = least reactive):

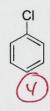






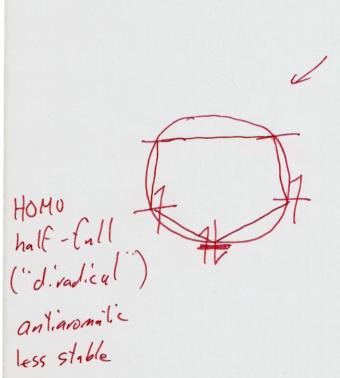






10

6. Give Molecular Orbital (MO) diagrams (hint: you can use a Frost circle) for the cyclopentadienyl cation and anion. Using them, show which is aromatic and which is antiaromatic. Also using the MO diagrams, explain why the aromatic electron configuration is more stable than the antiaromatic.



A A A

more stable

a)

- If stereo is issue, that's - 1/3
- wrong regionelectionty - 1
- minor product included with

= minor product included u.4

markoun. kou = 113 major =

b)

c)

d)

$$\frac{H_2SO_4}{\Delta}$$

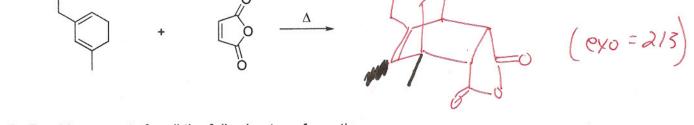
$$SO_3H$$

Wrong regischem

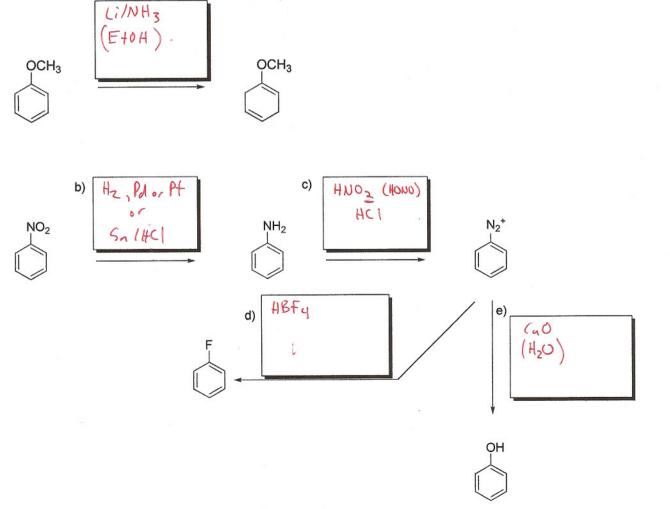
e)

f)

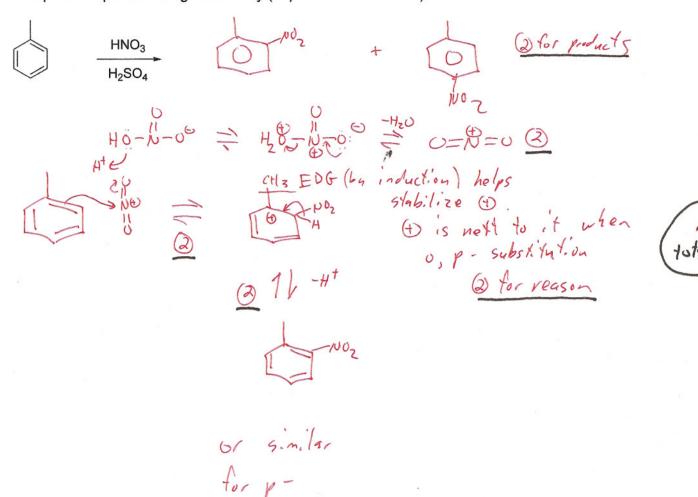
h)



- 8. Provide reagents for all the following transformations:
- a)



9. Provide the major product or products for the nitration of toluene, as well as a mechanism for the formation for one such product. The mechanism should include the generation of the electrophile as well as the addition of the electrophile to the aromatic ring. Use the mechanism to explain the predicted regionselectivity (*o-lp-* or *m-* substitution) for the reaction.



10. Provide a synthesis for one of the following compounds from benzene. Retrosynthetic analysis may also be provided for partial credit. If you show work on both, CLEARLY INDICATE WHICH SYNTHESIS YOU WANT GRADED. Otherwise, the one first worked on will be graded.