

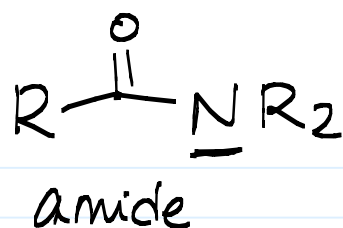
Carboxylic Acids & Derivatives (Ch 17 + 18)

Note Title

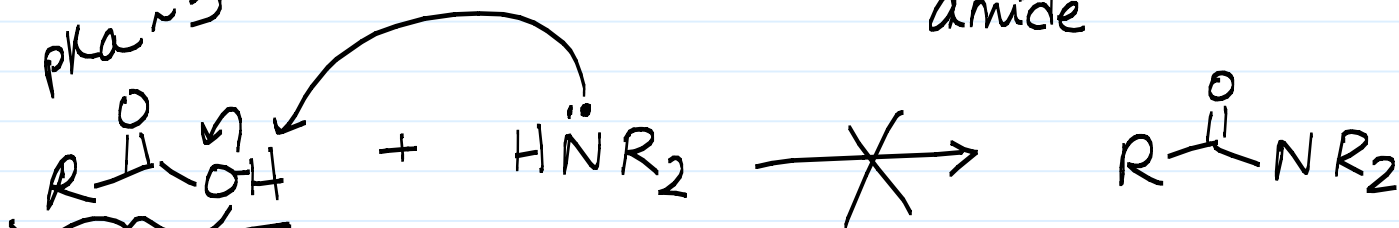
4/17/2014

Announcement: Get Midterm 1 after class.

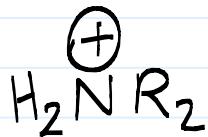
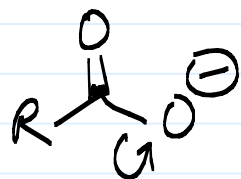
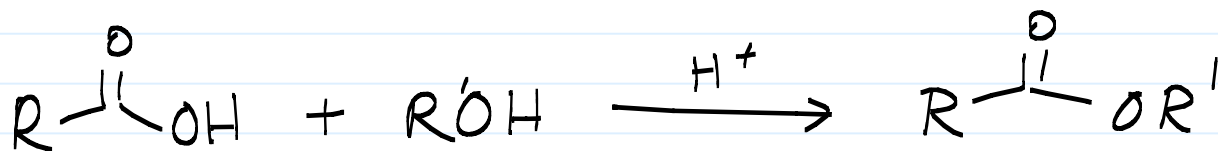
Amide Formation



pKa ~ 5



Compare to



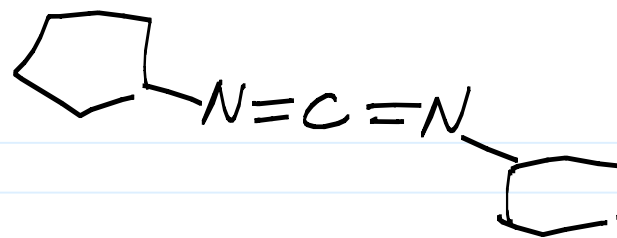
pKa = 9

O²⁻

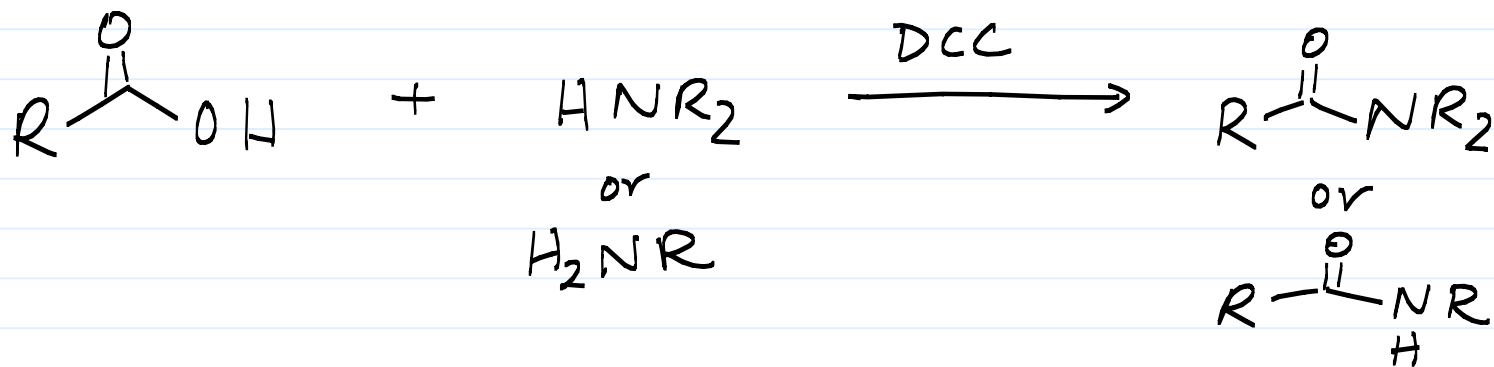
← HORRIBLE LEAVING GROUP!!

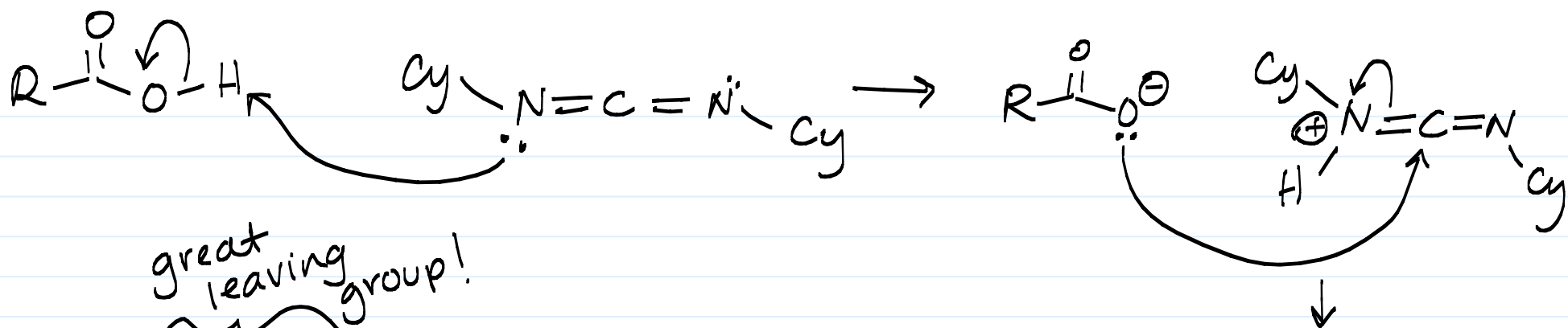
Dicyclohexylcarbodiimide

DCC

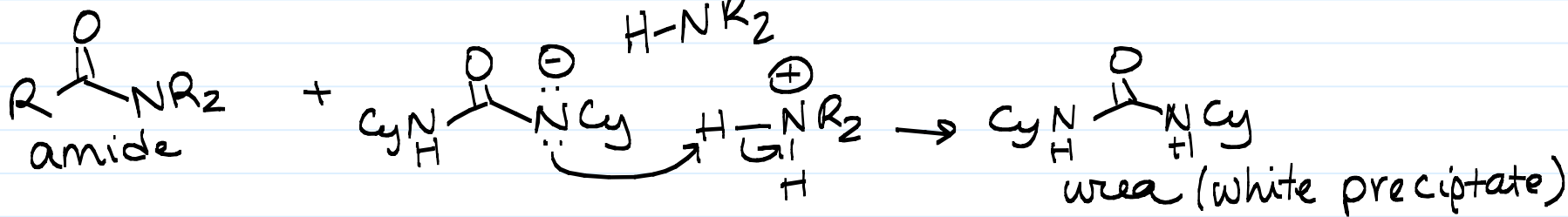
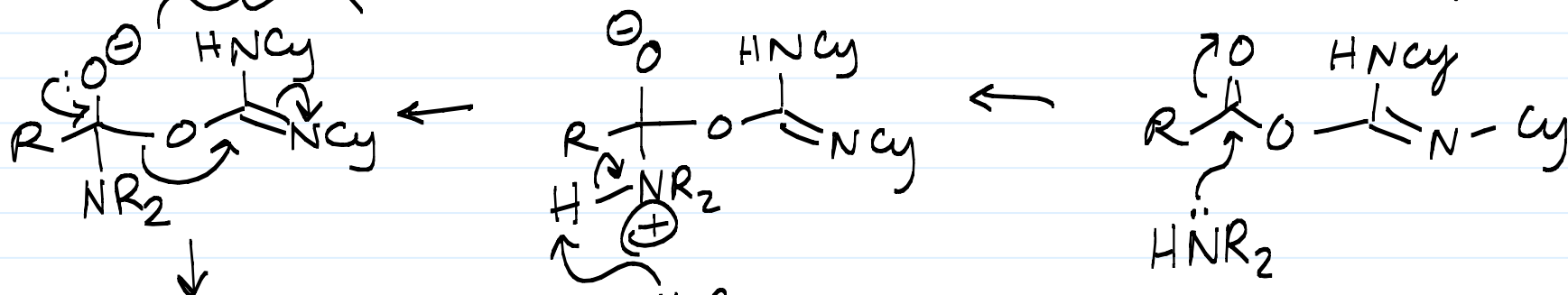


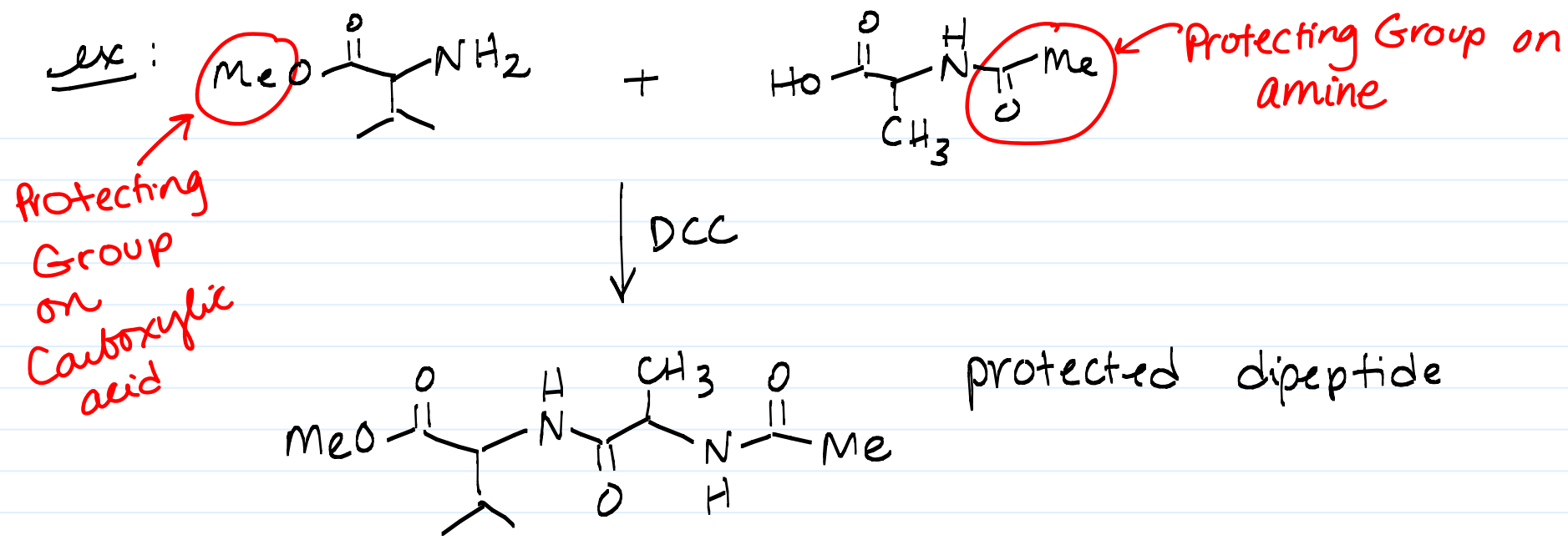
DCC Coupling to Form Amides





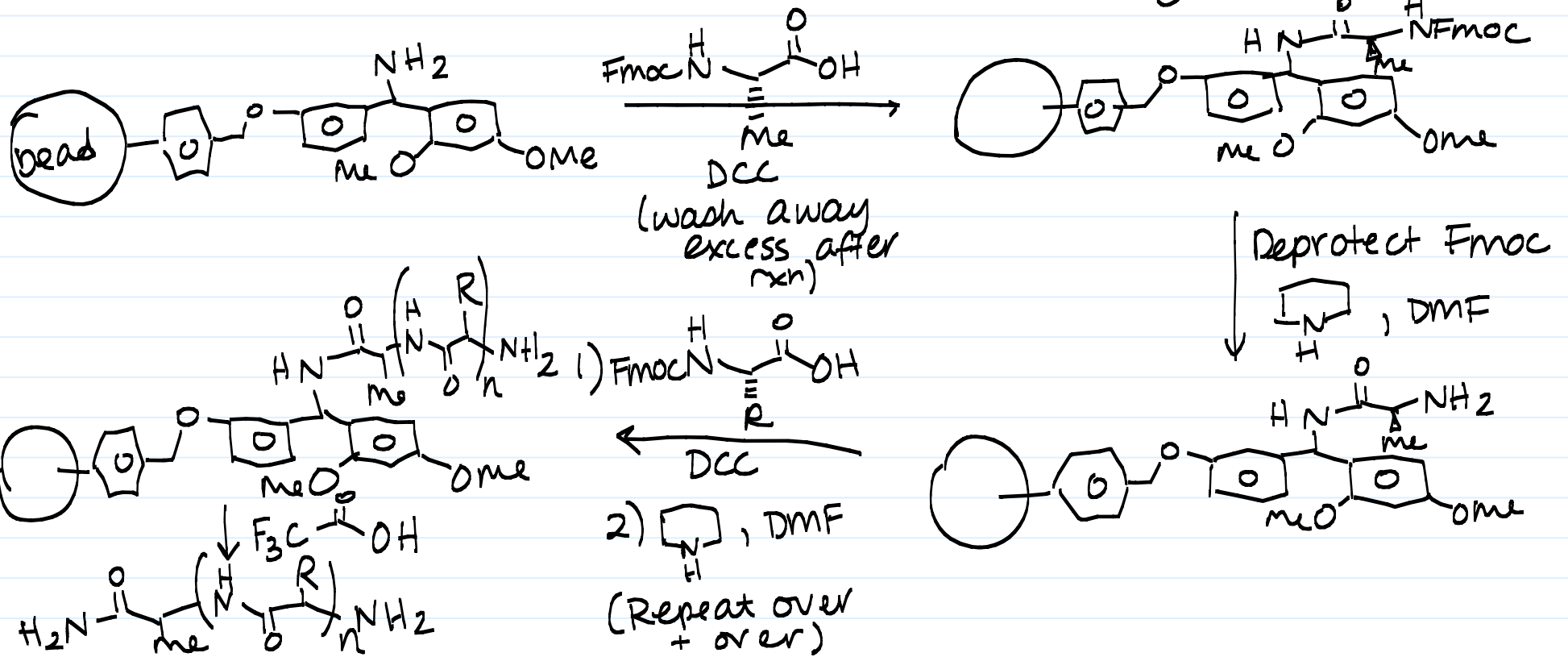
great leaving group!

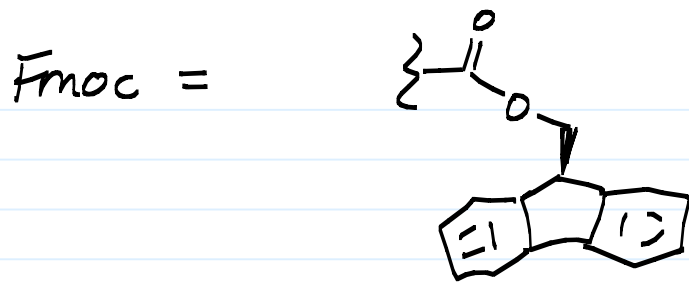




DCC couplings are efficient enough that you can do Solid Phase Peptide Synthesis (SPPS)... and a machine can even do it for you!!

Solid Phase Peptide Synthesis (SPPS) - Pioneered by Bruce Merrifield

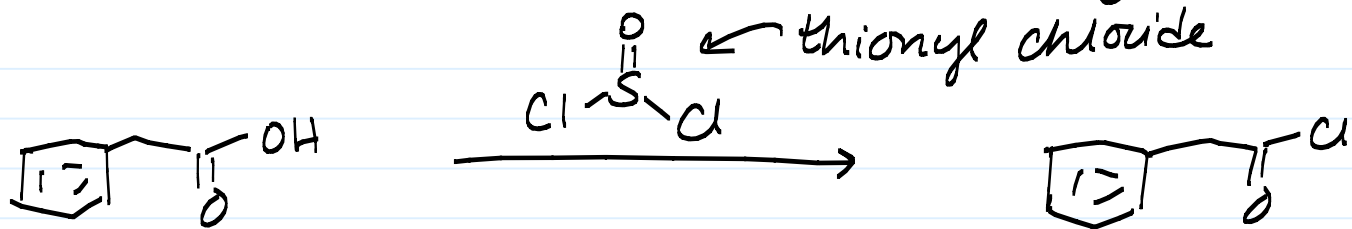




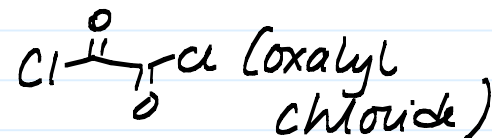
Note: $n \leq 70$. Making longer peptides with this technology is problematic... yields are too low for each step.

SPPS transformed the way that we make peptides... Changes what we can make + the types of questions we can ask.

Formation of Acid Chlorides (or Acyl chlorides)

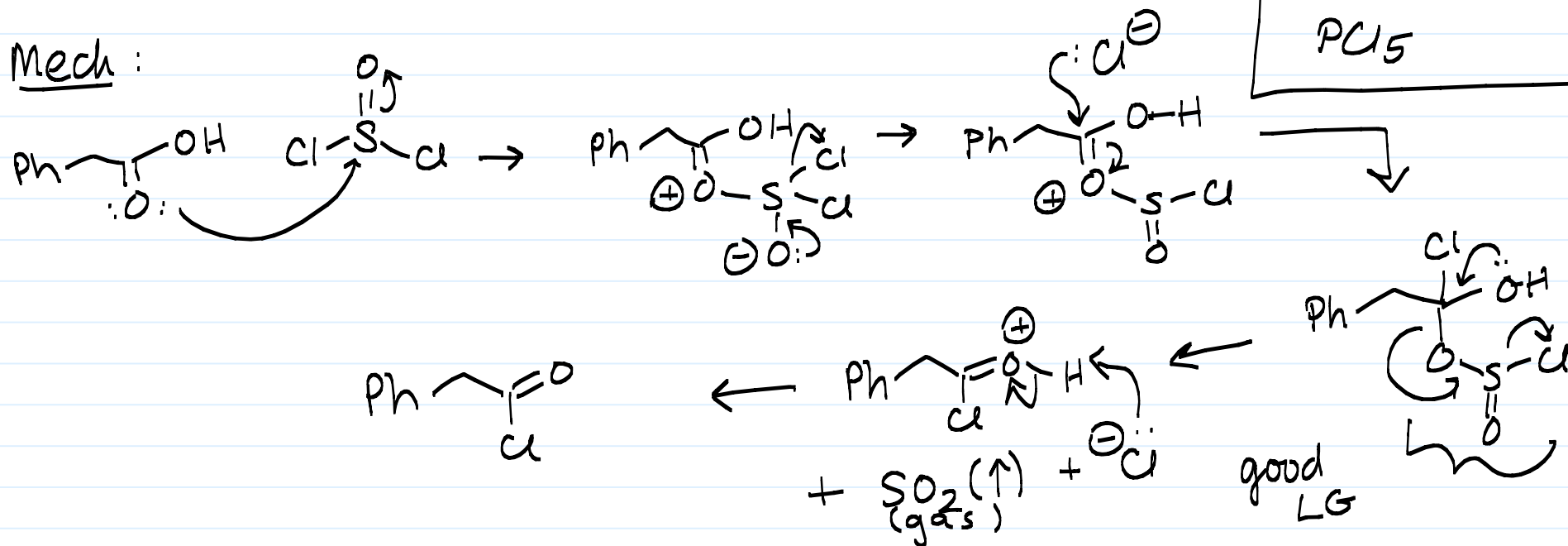


Other reagents you can use:



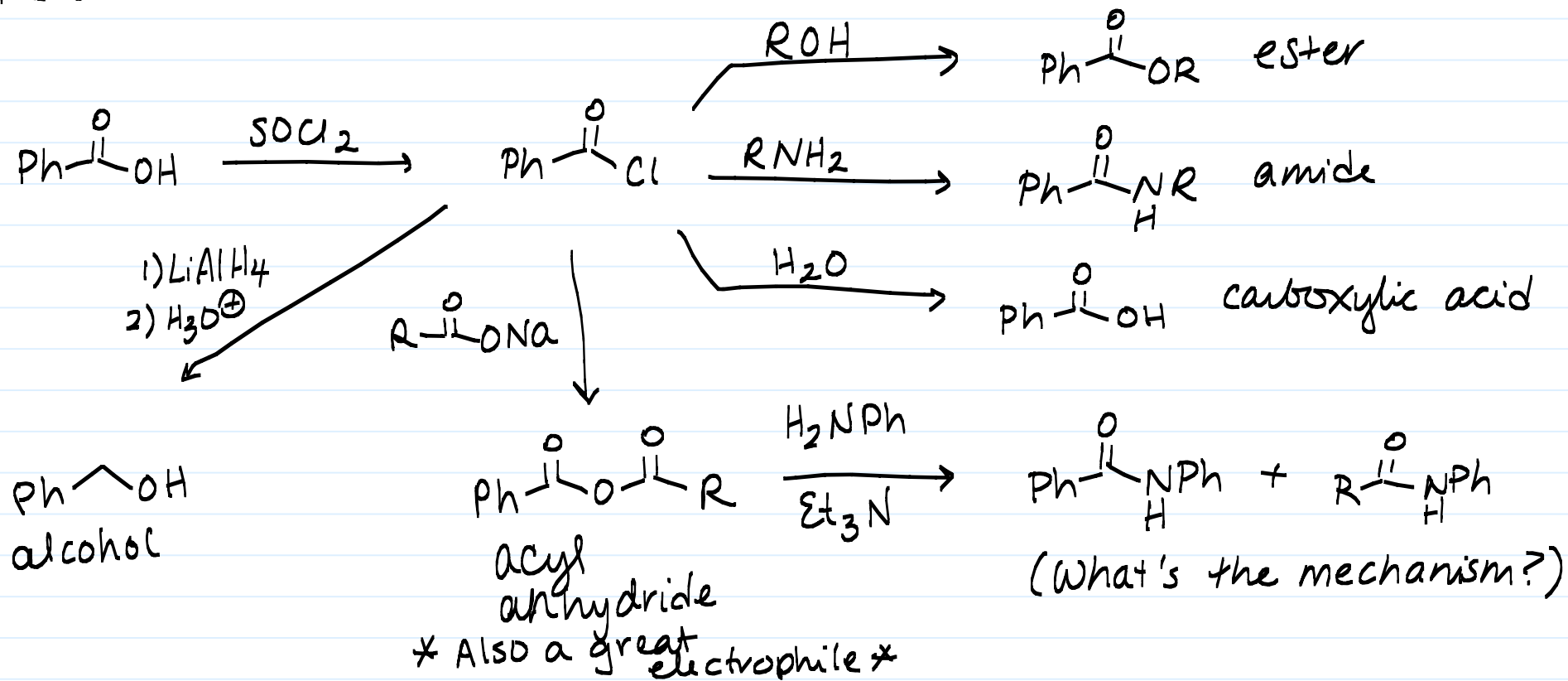
PCl5

Mech:

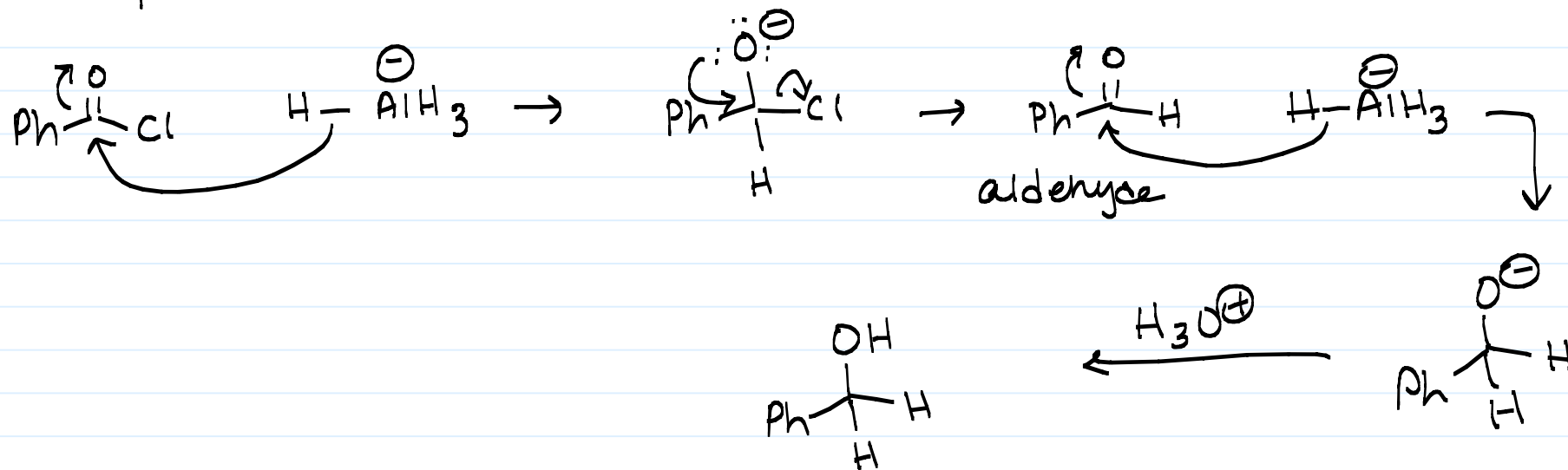


Reactions of Acid Chlorides

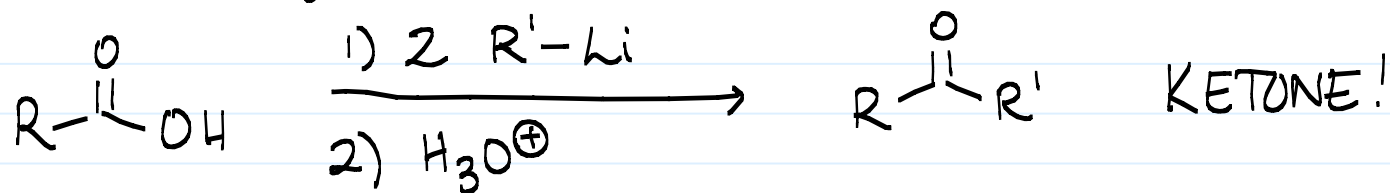
Recall: Acid chlorides in Electrophilic Aromatic Substitution Rxns.



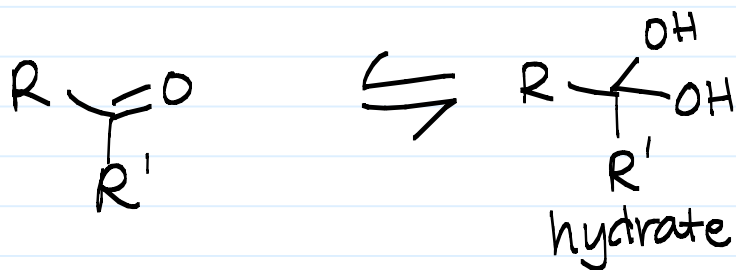
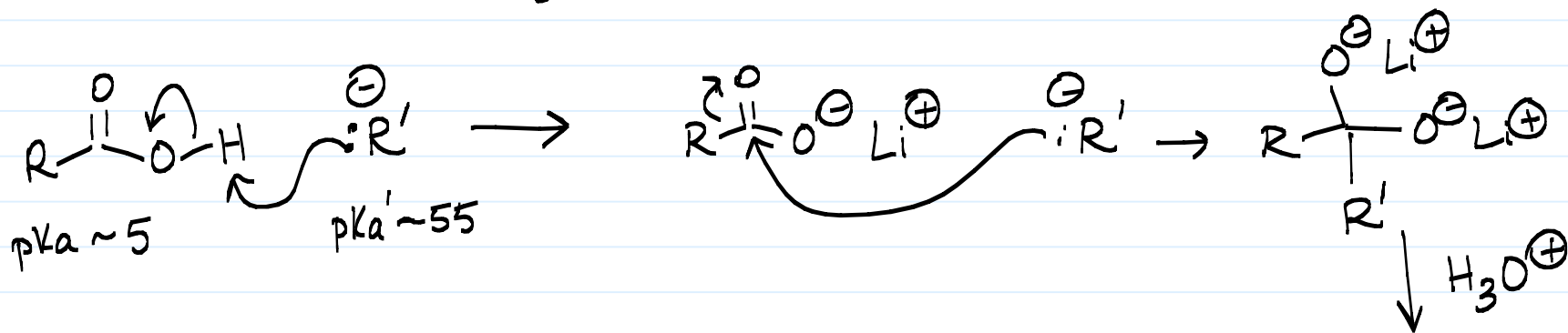
Mech of LiAlH_4 Rxn:



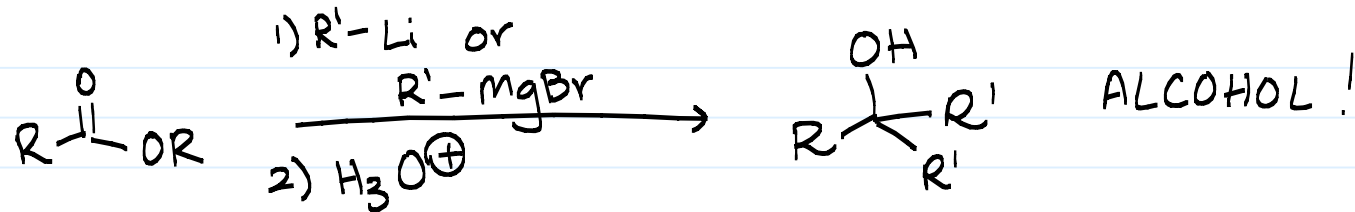
Reaction of Carboxylic Acids w/ RLi



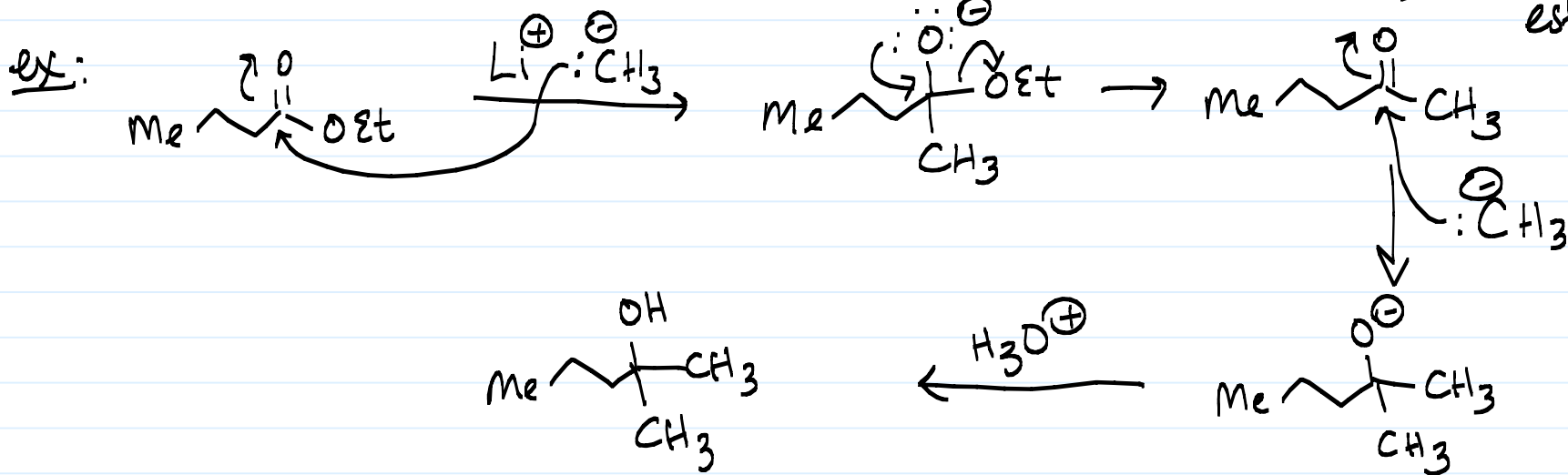
Mech:



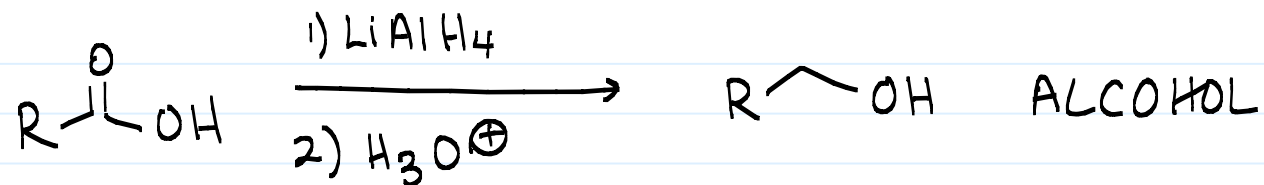
You can also add $R'-Li$ to esters, but you get a different product...



ketone is more electrophilic than ester, so rxn keeps going...



Rxns of Carboxylic Acids w/ LiAlH₄

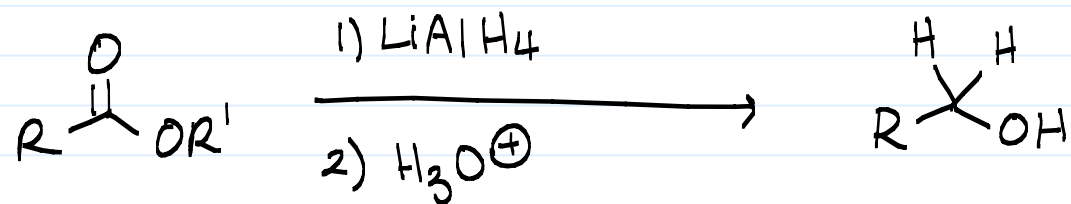


Mech is similar to rxns w/ RLi, but not fully known.

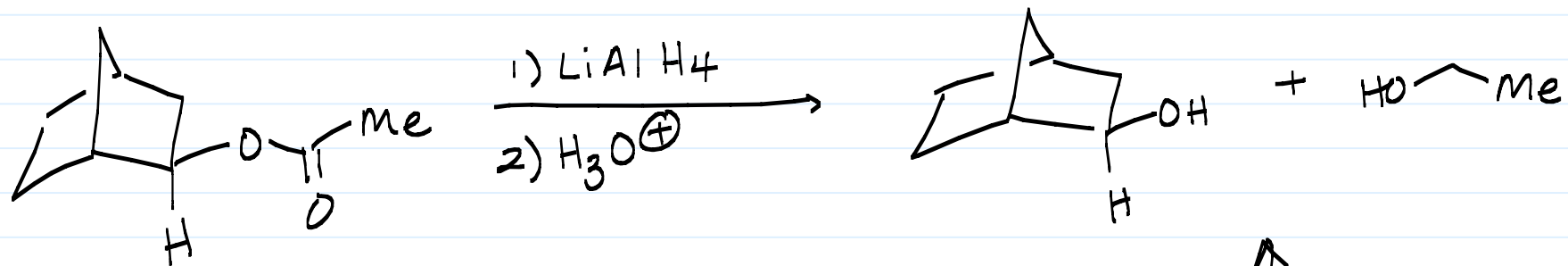
See p. 858 for our best guess.

Can also add LiAlH₄ to esters...

Esters + LiAlH₄



ex:



* Practice drawing the mechanism of this rxn *