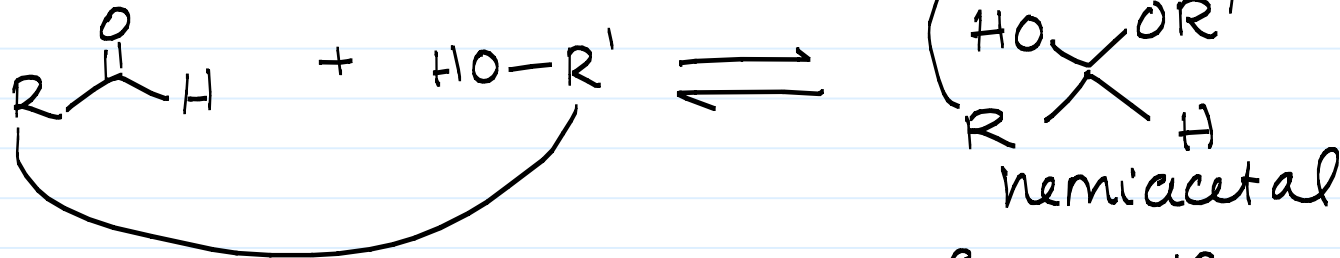


Sugars

Note Title

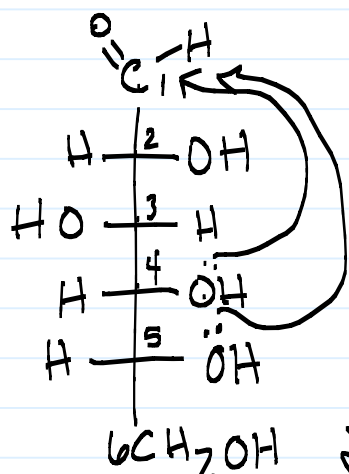
5/8/2014

Recall:

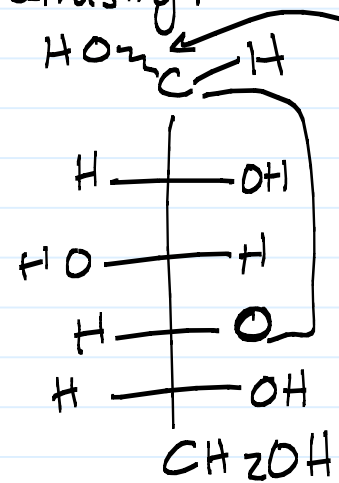


Really important in sugar chemistry!

→ favored if 5- or 6-membered ring can form.

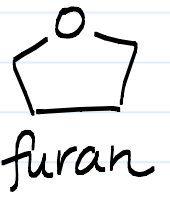


close @ C4

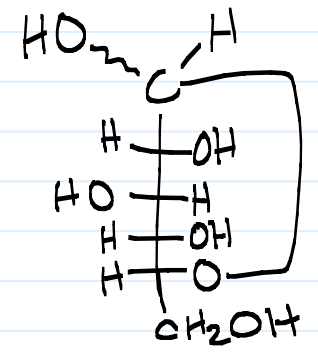


squiggle to indicate both configurations exist

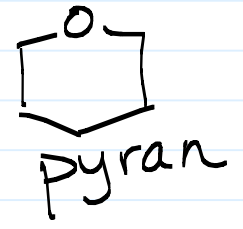
5-membered ring = "furanose"



close @ C5



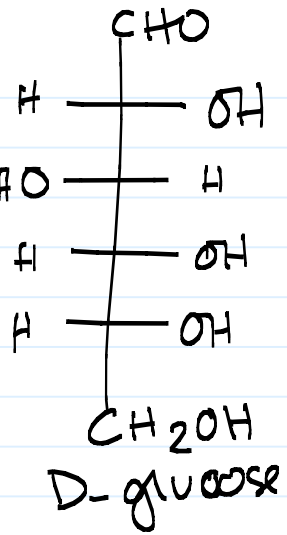
6-membered ring = "pyranose"



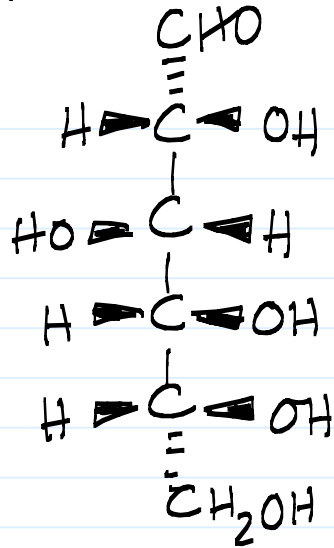
D-glucose

For glucose
Open < 0.05%
pyranose > 99.95%

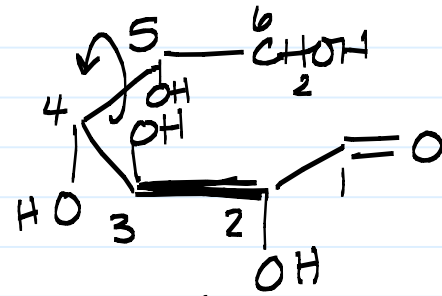
Haworth form:



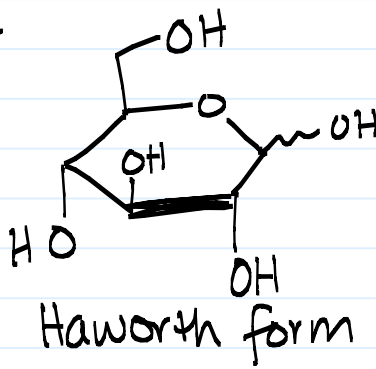
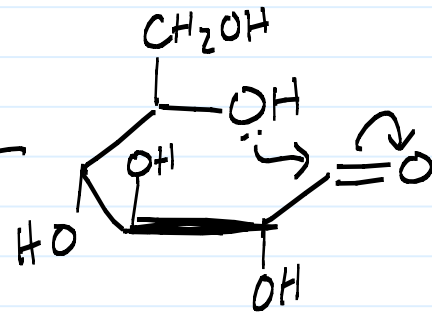
=



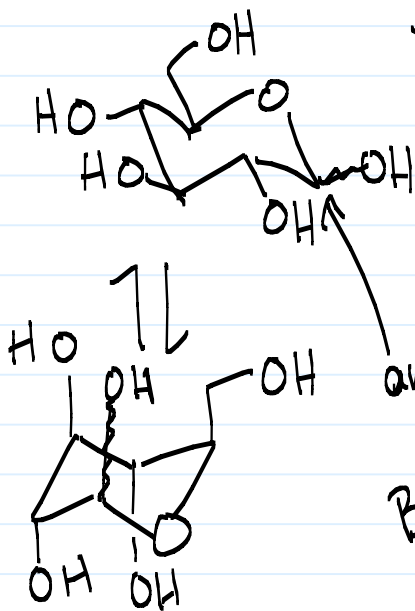
rotate +
wrap around
an imaginary
cylinder



Rotate around C4-C5



"relax" to chair

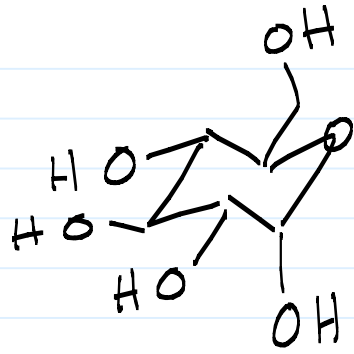


anomeric carbon.

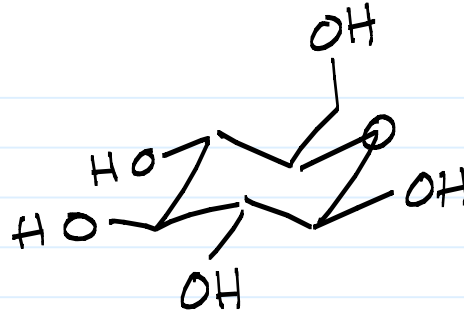
Both configurations exist (squiggle)

For glucose:

pyranose



α -glucose

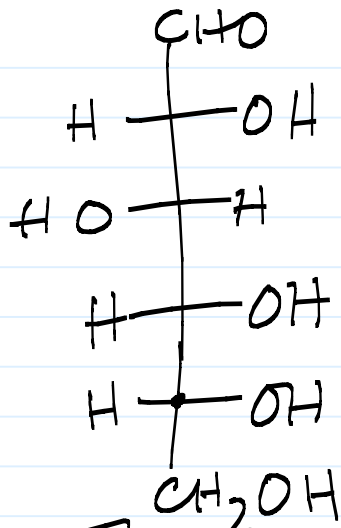


β -glucose

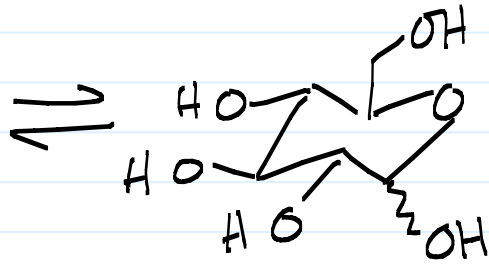
α -pyranose : β -pyranose : α -furanose : β -furanose
36 : 64 : <1 : <1

* Easiest to do with model kit!

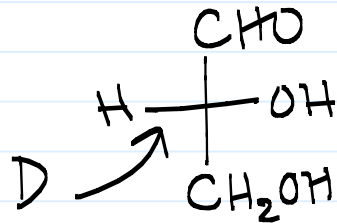
Sugars you should know by name:



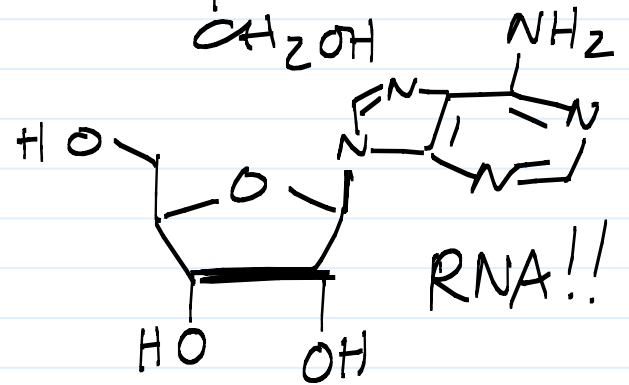
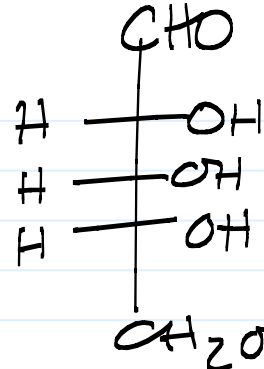
glucose



D vs. L glyceraldehyde



Ribose



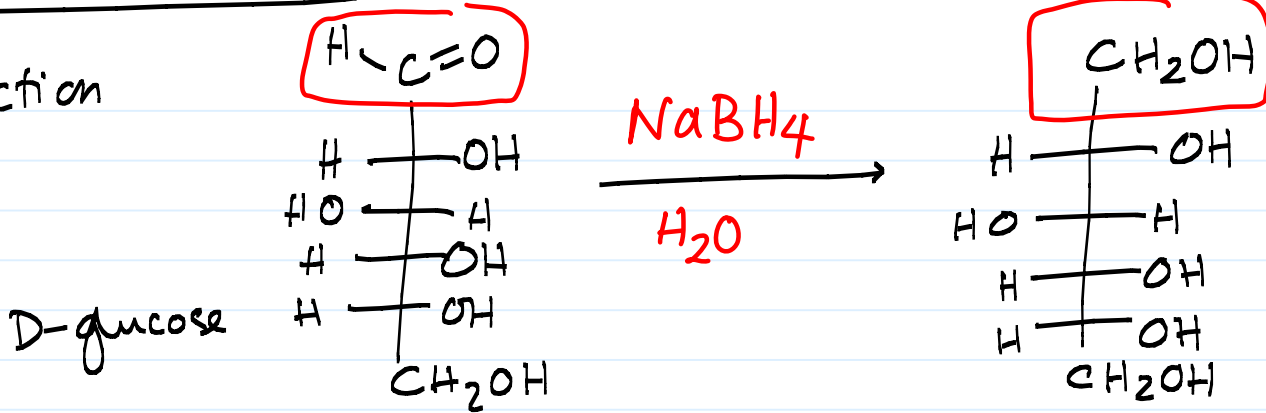
RNA!!

Dihydroxyacetone



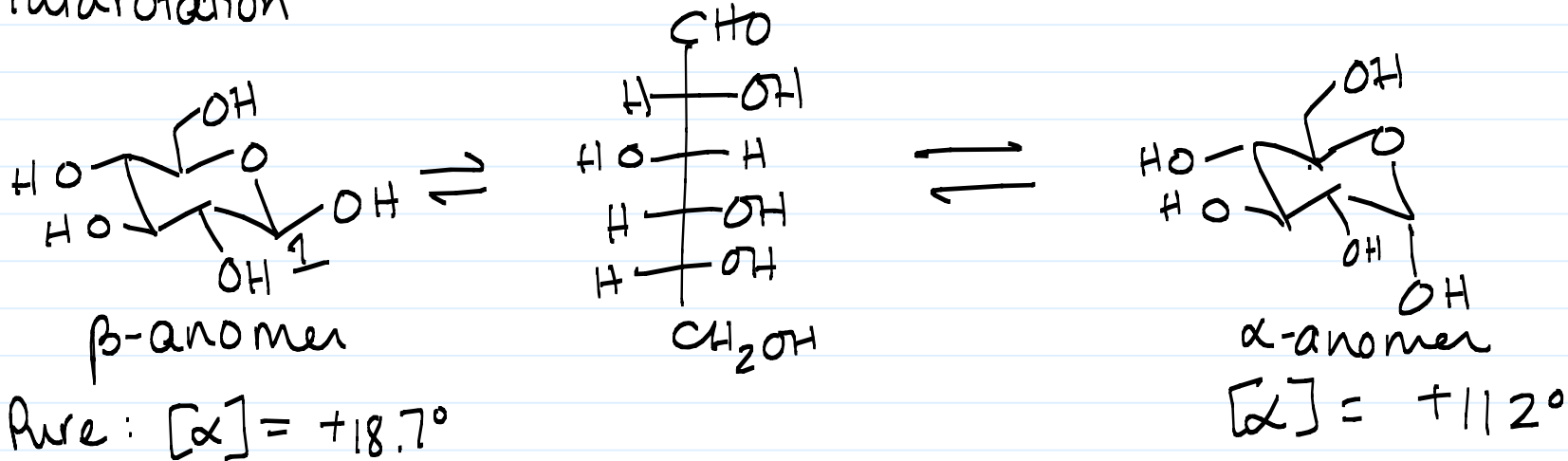
RXNS OF SUGARS

• Reduction



Note: NaBH₄ can reduce any aldehyde (more mild than LiAlH₄). Doesn't touch esters, carboxylic acids, etc.

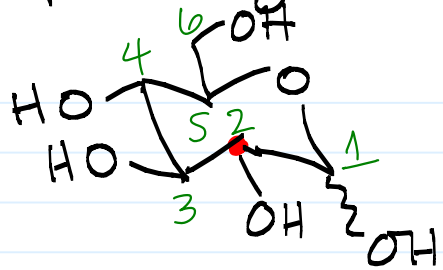
• Mutarotation



Aqueous Solution: $[\alpha] = +52.7^\circ$

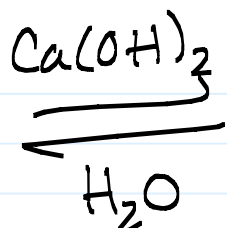
Practice: Draw mech!!

• Epimerization in Base

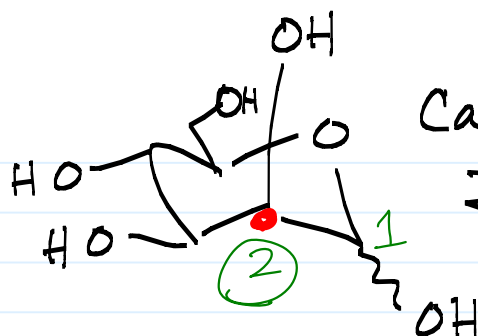


D-gucopyranose

63.4%

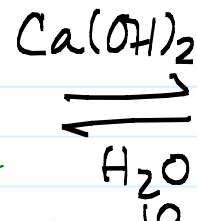


10 days

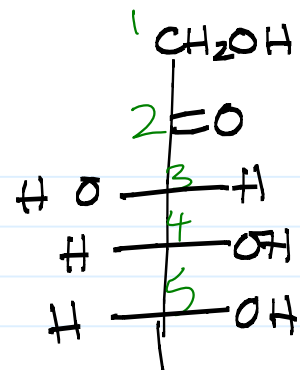


D-mannopyranose

2.4%

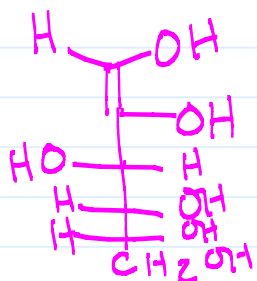
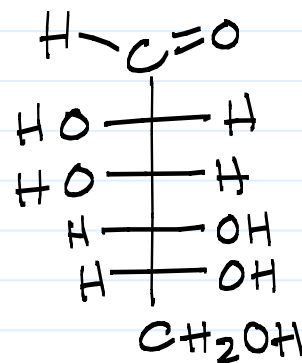
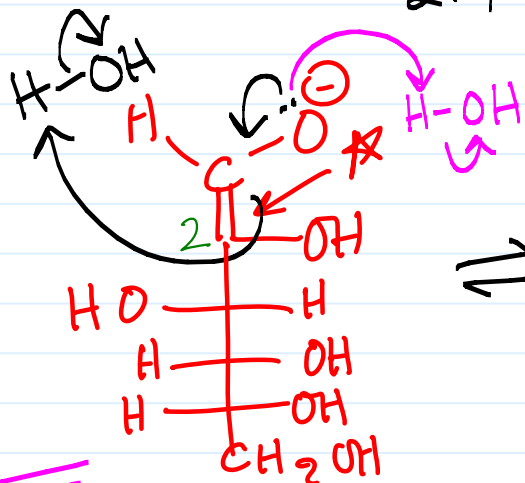
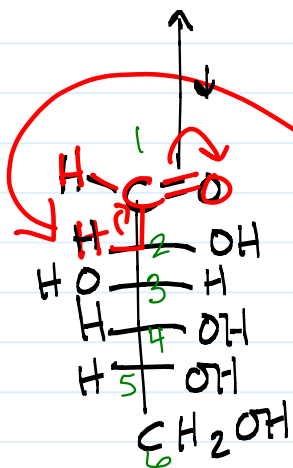


10 days

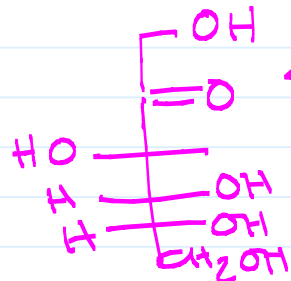


D-fructose

30.9%



tautomerize

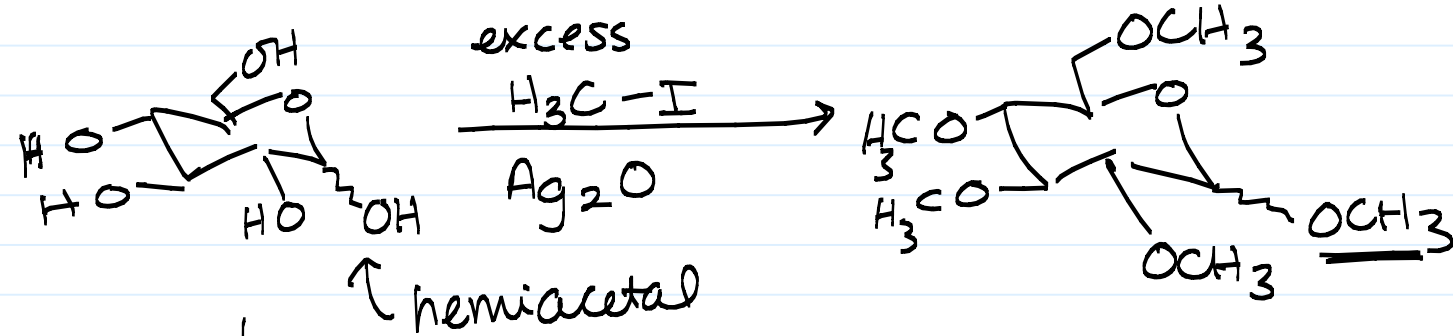


fructose

• Ether Form'n

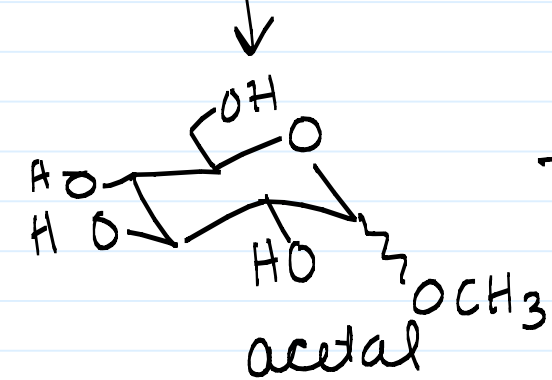


PRACTICE:
Draw mech's!!



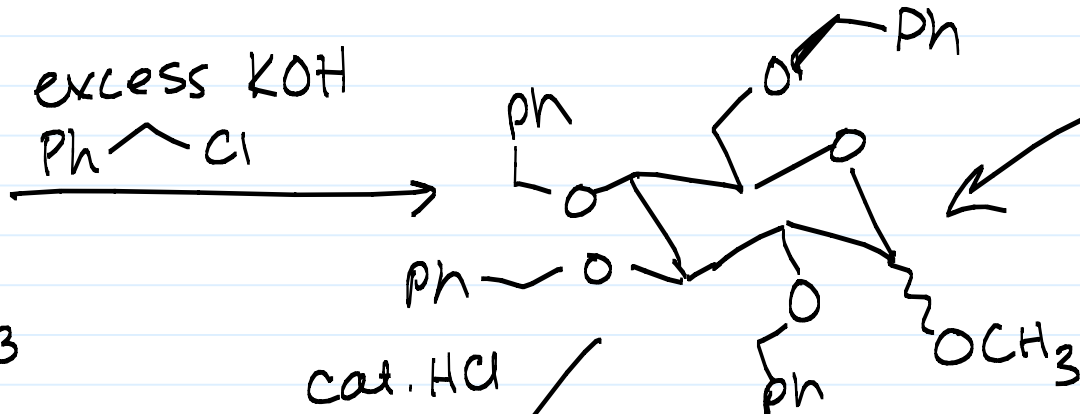
↑ hemiacetal

cat. HCl
H-OCH₃



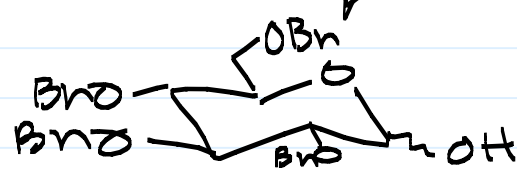
acetal

excess KOH
Ph-CH₂-Cl



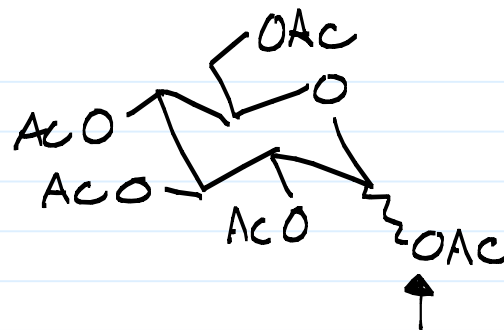
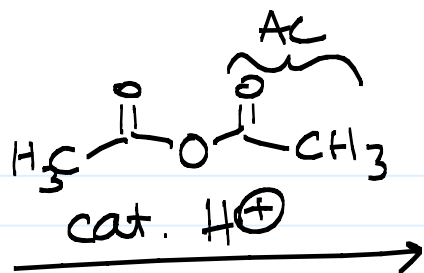
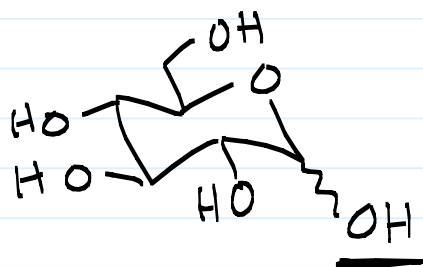
Acetal stable under basic conditions.

cat. HCl
H₂O



Bn = CH₂Ph

• Ester Formation



Draw mechanism!

↑
Where does this O
come from?